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$t = 2\,h$

$m = m_l = 5\,kg$

$t_0 = 5\,°C$

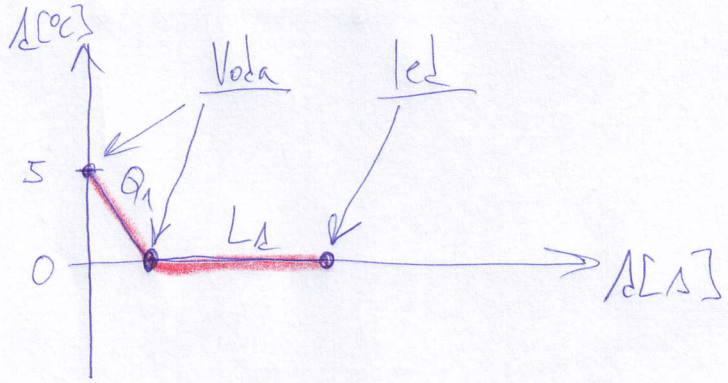
$t = 0\,°C$

$Q = ?\,[J]$

$c_v = 4200\,\frac{J}{kg \cdot °C}$

$l_t = 334\,000\,\frac{J}{kg}$

Náčrt:



$Q = Q_1 + L_t$

$Q = 105\,000 + 1\,670\,000$

$Q = 1\,775\,000\,J$

$Q \doteq 1{,}8\,MJ$

$Q_1 = m \cdot c_v \cdot (t_0 - t)$

$Q_1 = 5 \cdot 4200 \cdot (5-0)$

$Q_1 = 105\,000\,J$

$L_t = m \cdot l_t$

$L_t = 5 \cdot 334\,000$

$L_t = 1\,670\,000\,J$

Vodě bylo odebráno 1,8 MJ tepla.